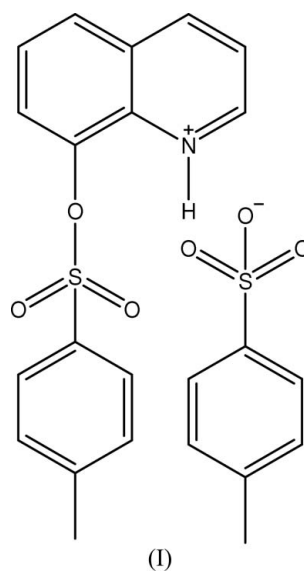


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Key indicators

Single-crystal X-ray study
 $T = 292$ K
Mean $\sigma(\text{C}-\text{C}) = 0.004$ Å
 R factor = 0.042
 wR factor = 0.120
Data-to-parameter ratio = 14.1For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.8-(Tolyl-4-sulfonyloxy)quinolinium
toluene-4-sulfonateIn the cation of the title compound, $\text{C}_{16}\text{H}_{14}\text{NO}_3\text{S}^+ \cdot \text{C}_7\text{H}_7\text{O}_3\text{S}^-$,
the quinolinium ring system makes a dihedral angle of
 $69.14(9)^\circ$ with the benzene ring.Received 4 November 2006
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Comment

Derivatives of 8-hydroxyquinoline are known for their anti-
amoebic, antibacterial and antifungal activities (Balasu-
bramanian & Thomas Muthiah, 1996; Khan *et al.*, 1994; Vembu
et al., 2003). The title compound, (I), is an intermediate in the
synthesis of 8-hydroxyquinoline derivatives (Lee *et al.*, 2001).
We report here the crystal structure of (I) (Fig. 1).The benzene C10–C15 and C22–C17 planes make dihedral
angles of $69.14(9)$ and $78.31(6)^\circ$, respectively, with the C1–
C9/N1 plane. The dihedral angle between the two benzene
rings is $61.74(9)^\circ$.In the crystal structure of (I), there are C–H...O and N–
H...O hydrogen bonds (Table 1).

Experimental

A solution of 8-hydroxyquinoline (6.9 mmol) in dichloromethane
(6 ml) was added dropwise to 4-toluenesulfonyl chloride (20.5 mmol)
and the mixture was refluxed in air for 4 h. The resultant precipitate
was filtered off and the solution was evaporated to give (I) as a deep-
yellow solid. Single crystals suitable for X-ray diffraction were
obtained by recrystallization from an acetone solution.

Crystal data

$C_{16}H_{14}NO_3S^+ \cdot C_7H_7O_3S^-$
 $M_r = 471.53$
 Triclinic, $P\bar{1}$
 $a = 9.643 (5) \text{ \AA}$
 $b = 10.852 (5) \text{ \AA}$
 $c = 11.421 (5) \text{ \AA}$
 $\alpha = 105.470 (5)^\circ$
 $\beta = 97.976 (5)^\circ$
 $\gamma = 99.295 (5)^\circ$

$V = 1115.8 (9) \text{ \AA}^3$
 $Z = 2$
 $D_x = 1.404 \text{ Mg m}^{-3}$
 Mo $K\alpha$ radiation
 $\mu = 0.28 \text{ mm}^{-1}$
 $T = 292 (2) \text{ K}$
 Block, yellow
 $0.45 \times 0.26 \times 0.24 \text{ mm}$

Data collection

Bruker SMART APEXII CCD
 area-detector diffractometer
 φ and ω scans
 Absorption correction: multi-scan
 (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.886, T_{\max} = 0.936$

8405 measured reflections
 4111 independent reflections
 3208 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.019$
 $\theta_{\text{max}} = 25.5^\circ$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.042$
 $wR(F^2) = 0.120$
 $S = 1.03$
 4111 reflections
 291 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0577P)^2 + 0.4629P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.001$
 $\Delta\rho_{\text{max}} = 0.35 \text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.29 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry ($\text{\AA}, ^\circ$).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$N1-H1 \cdots O4^i$	0.86	1.89	2.703 (3)	158
$C4-H4 \cdots O3^{ii}$	0.93	2.33	3.256 (3)	173
$C8-H8 \cdots O5^i$	0.93	2.54	3.358 (4)	146
$C15-H15 \cdots O4^i$	0.93	2.49	3.162 (3)	130
$C18-H18 \cdots O5^i$	0.93	2.59	3.288 (3)	132

Symmetry codes: (i) $-x + 1, -y + 1, -z + 2$; (ii) $x - 1, y, z$.

H atoms were positioned geometrically, with $C-H = 0.93-0.96 \text{ \AA}$, and refined as riding, with $U_{\text{iso}}(H) = 1.2U_{\text{eq}}(C)$ or $1.5U_{\text{eq}}(\text{methyl } C)$.

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997a); molecular graphics:

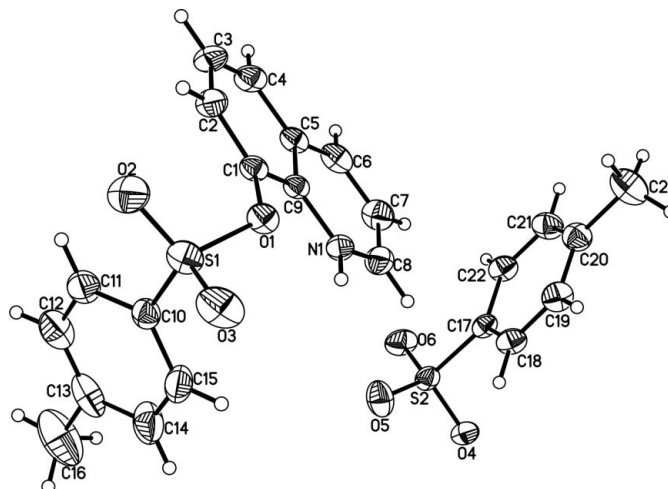


Figure 1

The molecular structure of (I), showing 30% probability displacement ellipsoids and the atomic numbering.

SHELXTL (Sheldrick, 1997b); software used to prepare material for publication: SHELXTL.

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